

The interaction between e_A of the prismane part and π^* of the CO group causes an electron transfer from the prismane, especially the upper three-membered ring moiety, to the CO group. As a consequence the antibonding character of C(1)–C(6) and the bonding character of C(1)–C(2) and C(2)–C(6), respectively, are reduced. This leads to the bond alternation for (3).

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Structure and Stereochemistry of 2-(*N*-Methylaniline)-3-(*N*-methoxyaceto)-7-(*p*-chlorophenyl)-8-methoxy-4,5-benzo-3-aza-1-nonem*†

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Abstract. $C_{28}H_{26}ClN_3O_4$, $M_r = 504.0$, monoclinic, $P2_1/n$, $a = 7.468$ (3), $b = 33.316$ (8), $c = 10.467$ (4) Å, $\beta = 100.28$ (3)°, $Z = 4$, $D_x = 1.31$ Mg m⁻³, Mo $K\alpha$, $\lambda = 0.7107$ Å, $\mu = 0.184$ mm⁻¹, $F(000) = 1056$, $T = 293$ K, final $R = 0.075$ for 2003 reflections. The

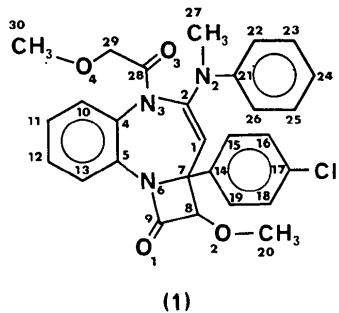
structure established chemically is confirmed. The seven-membered ring adopts a boat-shaped conformation. The dihedral angles between the best planes through the methoxyacetyl group and the phenyl ring of the *N*-methylanilino group at C(2), and between the *p*-chlorophenyl ring and methoxy group at C(8) are 128.0 (8) and 53.6 (8)°, respectively, and thus reduce the crowding of these bulky groups on the diazepine ring. The packing in the crystal is entirely due to van der Waals forces.

* 2a-(*p*-Chlorophenyl)-2-methoxy-5-(2-methoxyacetyl)-4-(*N*-methylanilino)-1,2,2a,5-tetrahydroazeto[1,2-a][1,5]benzodiazepin-1-one.

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Introduction. A great number of studies have been devoted to the search for structural analogues of cephalosporin and penicillin with specific activity (Guthikonda, Cama & Christensen, 1974; Bose, Ram, Hoffman, Hutchinson & Manhas, 1979; Itil, Saletu & Marasa, 1974). As part of our studies (Cortés & Martínez, 1983) on the total synthesis of new antibiotics that combine the two functionalities of cephalosporin and penicillin, in which the β -lactam moiety is integrated into the benzodiazepine system, and to investigate their pharmacological profile, synthesis of the title compound (1) (Cortés, Romero & Ceballos, 1986) was performed. We have undertaken the X-ray structure analysis of (1) in order to establish its molecular structure and stereochemistry.



Experimental. Single crystals of (1) were prepared and crystallized by Cortés *et al.* (1986). Prismatic crystal, 0.10 × 0.30 × 0.34 mm. Nicolet R3 four-circle diffractometer, graphite-monochromated Mo K α , lattice parameters from 24 machine-centred reflections with $4.6 < 2\theta < 13.7^\circ$. 5858 measured reflections with $3 < 2\theta < 55^\circ$, 2003 independent with $I > 3.0\sigma(I)$, index range $h \pm 8$, $k 0 \rightarrow 42$, $l 0 \rightarrow 13$, ω -scan mode, variable scan speed, scan width 1.0 (θ°). Two standard reflections (021, 170) monitored every 50 measurements throughout data collection process showed no significant changes in intensity. Intensities corrected for Lorentz–polarization factor but not for absorption. Data adjusted to an approximately absolute scale and an overall U value of 0.042 Å². Structure solved by direct methods and refined by least squares using a cascade-matrix procedure; coordinates and anisotropic temperature factors refined for the non-H atoms; H atoms in calculated positions with fixed isotropic $U = 0.06$ Å². Final $R = 0.075$, $wR = 0.086$. $\sum w(\Delta F)^2$ minimized; $w^{-1} = |\sigma^2(F_o) + 0.002(F_o)^2|$, where σ is the standard deviation of the observed amplitudes, based on counting statistics. In the last cycle $(\Delta/\sigma)_{\max} = 0.147$. Residual electron density in final difference map within ± 0.4 e Å⁻³. Atoms O(4), C(20) and C(30) of peripheral methoxy groups may be affected by some disorder (see thermal parameters in Table 1), giving rise to high values of R and wR . Atomic scattering factors from International Tables for X-ray Crystallography

(1974); isotropic extinction parameter $X = 0.00025$; $S = 1.236$. All computations performed on a Nova 4S computer and plots drawn on a Tektronix plotter with the SHELXTL system of programs (Sheldrick, 1981).

Discussion. The shape of the molecule and the atomic numbering are shown in Fig. 1. Final atomic coordinates are given in Table 1* and bond distances in Table 2. Calculations of the least-squares planes passing through the diazepine, benzo, and four-membered β -lactam rings and the two phenyl rings and the deviations of individual atoms from the planes and interplanar angles have been deposited.

The molecule consists of a benzodiazepine ring with methoxyacetyl, *N*-methylanilino and *p*-chlorophenyl groups with different orientations and a four-membered β -lactam ring with a methoxy group.

The conformation of the diazepine ring may best be seen in Fig. 1. It can be described as approximately boat-shaped; maximum deviations from the best plane are –0.495 (8), –0.407 (8) and 0.353 (8) Å for N(3), C(7) and C(1), respectively. This conformation is the same as that in 2-methylthio-7-(*p*-bromophenyl)-8-phenoxy-4,5-benzo-3-aza-2-nonem (Soriano-García, Toscano, Cortés & Martínez, 1984) and in classical benzodiazepines (Hamor & Martin, 1983; Butcher & Hamor, 1985). The angle between the mean planes of phenyl rings C(5)–C(4)–C(10)–C(11)–C(12)–C(13) and C(14) through C(19) is 93.7 (8)°. Similar interplanar angles occur in 2-methylthio-7-(*p*-bromophenyl)-8-phenoxy-4,5-benzo-3-aza-2-nonem (Soriano-

* Lists of structure amplitudes, anisotropic thermal parameters, H-atom coordinates, bond angles and the results of the mean-plane calculations have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 43332 (17 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

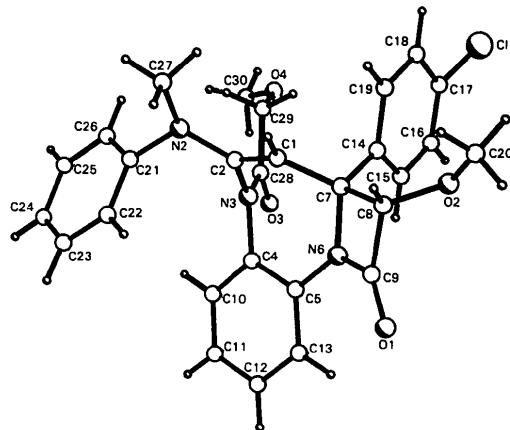


Fig. 1. View of molecule (1), showing the atom labelling.

García *et al.*, 1984) with a value of 85.9 (4)° and in clonazepam (Chananont, Hamor & Martin, 1979) where the halogen substituent is in the *ortho* position, with values of 83.7 and 77.8° for the two independent molecules.

The bond lengths C(5)–N(6) and N(6)–C(9) show the influence of conjugation between the keto group at C(9) and the phenyl ring C(5)–C(4)–C(10)–C(11)–C(12)–C(13). The considerable double-bond character in the C(5)–N(6) and N(6)–C(9) bonds is indicated by

Table 1. Atomic coordinates ($\times 10^4$) and equivalent isotropic temperature factors ($\text{\AA}^2 \times 10^3$)

$$U_{\text{eq}} = (U_{11} U_{22} U_{33})^{1/3}$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
Cl	464 (3)	-665 (1)	7593 (2)	78 (1)
N(2)	112 (6)	1563 (2)	9731 (5)	37 (2)
N(3)	1689 (6)	1330 (1)	8089 (4)	29 (2)
N(6)	-541 (7)	1289 (1)	5534 (5)	39 (2)
O(1)	-2283 (7)	1502 (2)	3540 (5)	75 (2)
O(2)	-3934 (6)	738 (1)	4791 (5)	68 (2)
O(3)	4350 (5)	990 (1)	8259 (5)	57 (2)
O(4)	3584 (9)	493 (2)	10143 (8)	124 (3)
C(1)	-1592 (8)	1309 (2)	7668 (6)	37 (2)
C(2)	-47 (8)	1407 (2)	8486 (6)	32 (2)
C(4)	2127 (8)	1556 (2)	7006 (6)	36 (2)
C(5)	1065 (8)	1527 (2)	5761 (6)	34 (2)
C(7)	-1506 (7)	1074 (2)	6460 (6)	35 (2)
C(8)	-3245 (9)	1103 (2)	5346 (7)	52 (3)
C(9)	-2052 (9)	1334 (2)	4571 (7)	51 (3)
C(10)	3646 (8)	1808 (2)	7218 (7)	47 (3)
C(11)	4110 (10)	2025 (2)	6194 (9)	62 (3)
C(12)	3133 (11)	1987 (2)	4958 (8)	62 (3)
C(13)	1608 (10)	1735 (2)	4738 (7)	55 (3)
C(14)	-907 (8)	639 (2)	6749 (5)	35 (2)
C(15)	523 (9)	462 (2)	6275 (7)	48 (3)
C(16)	983 (9)	61 (2)	6564 (7)	53 (3)
C(17)	1 (9)	-153 (2)	7307 (7)	50 (3)
C(18)	-1402 (10)	22 (2)	7819 (7)	54 (3)
C(19)	-1842 (9)	417 (2)	7535 (7)	47 (3)
C(20)	-5590 (11)	680 (4)	4640 (16)	266 (12)
C(21)	1525 (8)	1857 (2)	10142 (6)	39 (2)
C(22)	1594 (9)	2209 (2)	9450 (7)	46 (3)
C(23)	2987 (11)	2481 (2)	9848 (8)	60 (3)
C(24)	4304 (11)	2401 (3)	10891 (8)	65 (3)
C(25)	4268 (10)	2054 (3)	11582 (7)	62 (3)
C(26)	2843 (10)	1783 (2)	11206 (6)	53 (3)
C(27)	-1560 (9)	1607 (2)	10262 (7)	60 (3)
C(28)	2923 (7)	1033 (2)	8628 (6)	38 (2)
C(29)	2326 (8)	775 (2)	9660 (6)	43 (2)
C(30)	5062 (17)	546 (4)	10832 (13)	249 (11)

Table 2. Bond lengths (\AA) with e.s.d.'s in parentheses

Cl–C(17)	1.757 (7)	N(2)–C(2)	1.388 (8)
N(2)–C(21)	1.448 (8)	N(2)–C(27)	1.461 (9)
N(3)–C(2)	1.453 (8)	N(3)–C(4)	1.447 (8)
N(3)–C(28)	1.400 (7)	N(6)–C(5)	1.423 (7)
N(6)–C(7)	1.490 (8)	N(6)–C(9)	1.381 (8)
O(1)–C(9)	1.202 (9)	O(2)–C(8)	1.408 (8)
O(2)–C(20)	1.233 (10)	O(3)–C(28)	1.205 (8)
O(4)–C(29)	1.362 (9)	O(4)–C(30)	1.218 (14)
C(1)–C(2)	1.349 (8)	C(1)–C(7)	1.498 (9)
C(4)–C(5)	1.403 (8)	C(4)–C(10)	1.397 (8)
C(5)–C(13)	1.394 (10)	C(7)–C(8)	1.586 (8)
C(7)–C(14)	1.529 (8)	C(8)–C(9)	1.516 (11)
C(10)–C(11)	1.386 (12)	C(11)–C(12)	1.373 (12)
C(12)–C(13)	1.400 (11)	C(14)–C(15)	1.387 (9)
C(14)–C(19)	1.385 (9)	C(15)–C(16)	1.398 (9)
C(16)–C(17)	1.362 (10)	C(17)–C(18)	1.387 (11)
C(18)–C(19)	1.376 (10)	C(21)–C(22)	1.384 (9)
C(21)–C(26)	1.370 (8)	C(22)–C(23)	1.387 (10)
C(23)–C(24)	1.359 (11)	C(24)–C(25)	1.366 (12)
C(25)–C(26)	1.399 (11)	C(28)–C(29)	1.508 (9)

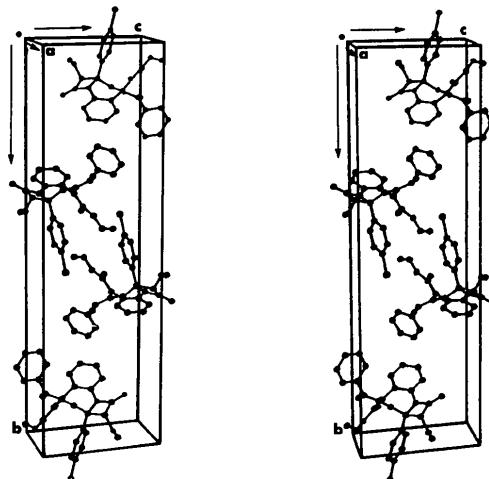


Fig. 2. Stereoscopic view of the unit cell.

the shortening of these bonds by 0.024 and 0.066 Å compared with the C(4)–N(3) bond length.

There is no conjugation between the seven-membered diazepine ring and the phenyl substituent at C(7). Owing to steric inhibition, caused by the neighbouring O(3) and O(4) of the methoxyacetyl group at N(3), and O(2) of the methoxy group at C(8), the *p*-chlorophenyl ring is rotated by an angle of 87.9 (7)° from the mean plane of the diazepine ring. The planes formed by the atoms of the methoxyacetyl group and the phenyl ring C(21) through C(26) make a dihedral angle of 128.0 (8)°. The phenyl rings show normal geometries, 1.380 Å and 120° for the mean C(sp²)–C(sp²) bond distance and the mean internal angle. This mean bond distance is close to that given by Sutton (1965) [1.395 (5) Å] and Brisse & Sygusch (1974) (1.379 Å).

The four-membered β -lactam ring and the three phenyl rings are planar to a good approximation.

The molecular packing is shown in Fig. 2. There are only four intermolecular approaches of non-hydrogen atoms < 3.41 Å, C(1)–O(3)(-1 + *x*, *y*, *z*) 3.371 (7); C(18)–O(4)(-*x*, -*y*, 2 - *z*) 3.375 (7) Å, and O(1)–Cl and C(16)–C(16)(-*x*, -*y*, 1 - *z*) 3.406 (7), 3.365 (7) Å, respectively. The molecules are, therefore, held in the crystal by van der Waals forces.

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Structure and Stereochemistry of 2-Hydroxy-12(S)-hydroxyneoclerodane-3,13(14)-diene-15,16;19,20-diolide (Semiatrin), a Diterpene*

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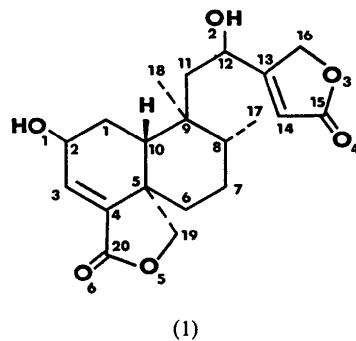
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Abstract. $C_{20}H_{26}O_6$, $M_r = 362.4$, orthorhombic, $P2_12_12_1$, $a = 8.564$ (3), $b = 10.550$ (4), $c = 20.456$ (5) Å, $V = 1848$ (1) Å³, $Z = 4$, $D_x = 1.30$ Mg m⁻³, Mo $K\alpha$, $\lambda = 0.7107$ Å, $\mu = 0.089$ mm⁻¹, $F(000) = 776$, $T = 293$ K, final $R = 0.061$ for 1284 reflections. The structure determination of (1) establishes the configuration at C(12) to be 12S. The cyclohexene and cyclohexane rings are *trans*-fused and adopt half-chair and chair conformations, respectively. The γ -lactone has an α -envelope conformation whilst the 13(16H)-furanone ring is planar to within ± 0.02 (1) Å. Bond lengths and angles are normal. The crystal structure is stabilized by intra- and intermolecular hydrogen bonds and C–H…O hydrogen-bond interactions.

Introduction. In a previous paper (Esquivel, Hernández, Ramamoorthy, Cárdenas & Rodríguez-Hahn, 1986) the isolation and structure determination of semiatrin (1) from *Salvia semiatrata* was described. The structure of (1) was assigned mainly on spectroscopic grounds. The *trans*-neoclerodane structure proposed for it and the 2S configuration at C(2) were deduced by careful analysis of the ¹H and ¹³C NMR spectra and comparison with data reported for similar structures. The configuration at C(12) could not be ascertained from the proton resonance data (Pinhey,

Simpson & Batey, 1971; Tschesche & Streuff, 1978; Esquivel, Méndez, Ortega, Soriano-García, Toscano & Rodríguez-Hahn, 1985; Bohlmann, Zdero & Huneck, 1985), although they suggested a 12S configuration. This prompted us to undertake the X-ray structure analysis of (1) in order to confirm the structure proposed for it and establish the configuration at C(12).



Experimental. Colourless transparent cube-shaped crystals prepared by slow evaporation of Me_2CO –hexane, dimensions 0.36 × 0.38 × 0.38 mm. Nicolet R3 four-circle diffractometer, graphite-monochromated Mo $K\alpha$, lattice parameters from 25 machine-centred reflections with $4.0 < 2\theta < 20.0^\circ$ accurately measured on the diffractometer. 2430 independent measured reflections, $2\theta_{\max} 55^\circ$, 1284 considered observed [$I >$

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